

Non-equilibrium thermodynamics

Exercise 4

Transport of heat and mass

1

- a) You have a system with 2 components dissolved in water and there is a temperature gradient and concentration gradients. Write the flux equations for the system.
- b) You have a system where the only driving force is a gradient in chemical potential for one of the components. Show how Fick's law can be derived from non-equilibrium thermodynamics. What is the relation between l_{ii} and Fick's diffusion coefficient for the component in the specific mixture?
- c) You have a system where the only driving force is a temperature difference. Show how Fourier's law can be derived from non-equilibrium thermodynamics. What is the relation between l_{qq} and the thermal conductivity of the material?
- d) What is the Soret effect?
- e) What is the definition of the thermal diffusion coefficient, D_T , the Soret coefficient, s_T , and the thermal diffusion factor, α_T ?
- f) What is the Dufour effect?
- g) What is the definition of the heat of transfer, q^* ?

2

A cylindrical container of length 5 mm is filled with an aqueous solution of KCl, $c_{KCl} = 0.01 \text{ kmol m}^{-3}$. The temperature on the left-hand side, l, is 20 °C, and on the right-hand side, r, 30 °C. In the stationary state a difference in concentration is established between the two sides, $c_{KCl,r} - c_{KCl,l} = \Delta c_{KCl} = -1.25 \cdot 10^{-4} \text{ kmol m}^{-3}$. The average Fick's diffusion coefficient for KCl in water is $D_{1,2} = 1.9 \cdot 10^{-9} \text{ m}^2\text{s}^{-1}$.

- a) Calculate the Soret coefficient, s_T , and the heat of transfer, q^* (use the mean temperature), for KCl in the solution.
- b) Calculate the flux of KCl at the starting time, when $\Delta c_{KCl} = 0$.

3

A cylindrical container of length 1 dm is filled with propane with some dissolved methane at 346 K and 56 bar. At the left side the concentration of methane is $3.7 \cdot 10^{-4} \text{ mol/m}^3$ and at the right side the concentration is $4.0 \cdot 10^{-4} \text{ mol/m}^3$. Assume that the diffusion coefficient of methane in propane is $6.2 \cdot 10^{-9} \text{ m}^2/\text{s}$ and that α_T is 8.9. Calculate the heat flux at these conditions.